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Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a. Na 2CO 3 2AgNO 3 → 2NaNO 3 3Ag 2CO 3(s) b. 2Na 3PO 4 3Pb(NO 3) 2 → 6NaNO 3 Pb 3(PO 4) 2(s) 2. Sodium hydroxide reacts with calcium chloride to form sodium chloride and solid calcium hydroxide. 3. Mixings d,n, and o all gave no visible

Chapter 11 Small-Scale Lab

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